

# Zeta potential and point of zero charge of clay fraction as affected by vinasse and phosphorus\*

Bruno Teixeira Ribeiro<sup>1</sup>, José Maria de Lima<sup>2</sup>, Nilton Curi<sup>2</sup>, Geraldo César de Oliveira<sup>2</sup> and Pedro Luiz Terra Lima<sup>2</sup>

<sup>1</sup> Agricultural Science Institute, Federal University of Uberlândia, Uberlândia, Minas Gerais State, Brazil. <sup>2</sup> Department of Soil Science, Federal University of Lavras, Lavras, Minas Gerais State, Brasil. Email: btribeiro@iciag.ufu.br; jmlima@dcs.ufla.br; niltcuri@dcs.ufla.br; geraldooliveira@dcs.ufla.br; pedroterralima@yahoo.com.br

\* Part of the doctorate thesis of the first author

## INTRODUCTION

### Soil charge - importance

✓ Soil dispersion and flocculation

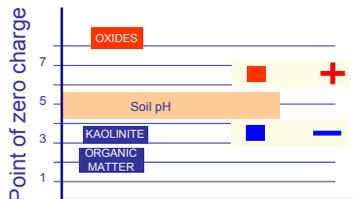
(Sumner, 1992; Chorom & Rengasamy, 1995)

✓ Cation and anion exchange (Gilman, 1985)

✓ Pesticides and heavy metals sorption

(Clausen et al., 2001; Appel & Ma, 2002)

### Tropical soils – variable charge (pH dependent)



### Brazil

➤ Ethanol production: 27 billion liters;

➤ Vinasse: the main byproduct of ethanol and brandy production. It is produced in large amounts - about 13 liters per liter of ethanol or brandy;

➤ Vinasse: 354 billion liters in 2009;

➤ Disposal in soils as liquid fertilizer is an alternative use for this product, mainly as a source of K. This practice avoids dumping vinasse into the water courses and lakes, which was common in the past (Günkel et al., 2007);

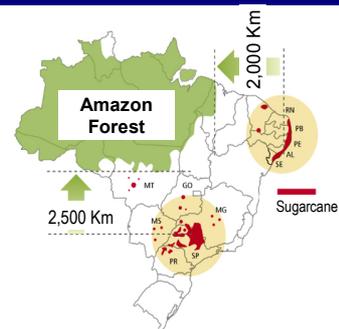
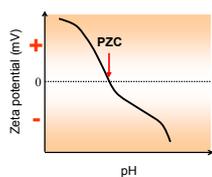
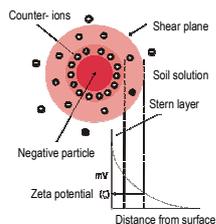


Figure 1. Distribution of sugarcane crops in Brazil. Extracted and adapted from [www.unica.com.br](http://www.unica.com.br)

### Zeta potential ( $\zeta$ )



- Zeta potential: the electrical potential developed at the solid-liquid interface (Tan, 1993)

- Point of zero charge (PZC): pH at which the surface charge is electrically neutral, for example, zeta potential = 0 (Tan, 1993)

## OBJECTIVES

The objective of this work was to evaluate the effect of vinasse, P sorption and the interaction vinasse-phosphorus on zeta potential and point of zero charge (PZC) of clay-fraction samples from two soils with different iron and aluminum contents (Red Latosol and Gleysol).

## MATERIAL AND METHODS

### - Clay fraction of Dystroferic Red Latosol (LVdf) and Gleysol (referenced as gibbsite)

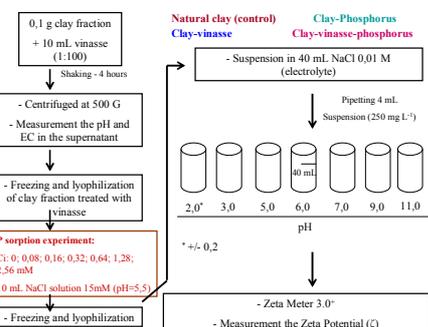
	SiO <sub>2</sub> <sup>1/</sup>	Al <sub>2</sub> O <sub>3</sub> <sup>1/</sup>	Fe <sub>2</sub> O <sub>3</sub> <sup>1/</sup>	Ki	Kr	TOC
	-g kg <sup>-1</sup>					-g kg <sup>-1</sup>
LVdf	284	256	177	1,88	1,31	6,0
Gibbsite	86	489	9,1	0,30	0,30	1,5

<sup>1/</sup> After acid sulfuric digestion (Embrapa, 1997); Ki: molecular ratio (SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>); Kr: molecular ratio (SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> + Fe<sub>2</sub>O<sub>3</sub>); TOC: total organic carbon.

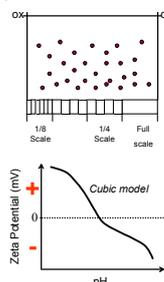
### - Vinasse from sugarcane spirit production

pH	CE	I	OC	P	K	Ca	Mg	S	Na	Cu	Mn	Zn	Fe
-----g L <sup>-1</sup> -----													
3,5	7,5	0,1	9,6	0,0	2,1	0,2	0,2	0,0	0,7	7,4	1,5	0,0	18,3
-----mg L <sup>-1</sup> -----													

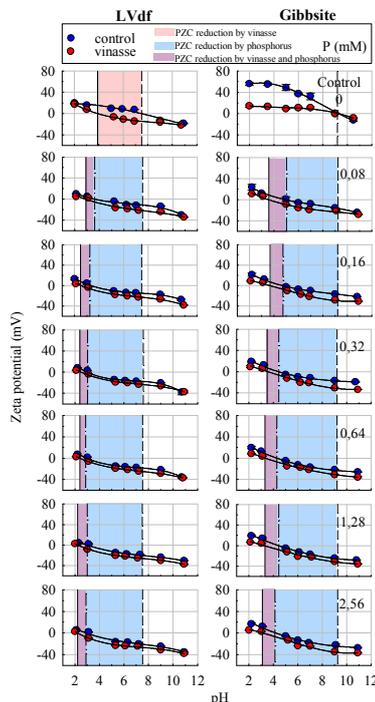
EC: electrical conductivity (dS m<sup>-1</sup>); I: ionic strength (mol L<sup>-1</sup>); I = CE.0,0127; OC: organic carbon.



• Counting: 10 to 20 particles for each replicate Voltage (50 – 100 V); "tracking" – 5s



## RESULTS



## CONCLUSIONS

✓ The vinasse influenced the zeta potential of clay fraction of both soils. In the Red Latosol, the PZC was reduced from 7.5 to 5.0, and the particles became more negatively charged. For gibbsite, the PZC was not changed, but the positive charges were significantly reduced. The P sorption increased negative charges to the soil particles and reduced the PZC. These effects were more pronounced when the clay fraction was previously treated with vinasse.

## ACKNOWLEDGMENTS

The authors show their appreciation to Fapemig, CNPq and Federal University of Uberlândia, Brazil, for the financial support to this research.

Figure 2. Zeta potential and point of zero charge (PZC) of Dystroferic Red Latosol and Gleysol (gibbsite) as affected by vinasse and phosphorus sorption. Erros bars indicate the standard deviation (n=3)

## REFERENCES

APPEL, C.; MA, L. Concentration, pH, and surface charge effects on cadmium and lead sorption in three tropical soils. *Journal Environmental Quality*, v. 31, p. 581-589, 2002.  
 CHOROM, M.; RENGASAMY, P. Dispersion and zeta potential of pure clays as related to net particle charge under varying pH, electrolyte concentration and cation type. *European Journal of Soil Science*, v. 46, p. 657-665, 1995.  
 CLAUSEN, L.; FABRICIUS, I.; MADSEN, L. Adsorption of pesticides onto quartz, calcite, kaolinite, and α-alumina. *Journal Environmental Quality*, v. 30, p. 846-857, 2001.  
 EMBRAPA. *Manual de Métodos de Análises de Solo*. 2 ed. Rio de Janeiro: Embrapa: Centro Nacional de Pesquisa de Solos, 1997. 212 p.  
 GILMAN, G.P. Influence of organic matter and phosphate content on the point of zero charge of variable charge components in oxidic soils. *Australian Journal of Soil Research*, v. 23, p. 643-646, 1985.  
 GÜNKEL, G.; KOSMOL, J.; SOBRAL, M.; ROHN, H.; MONTENEGRO, S.; AURELIANO, J. Sugar cane industry as a source of water pollution – case study on the situation in Ipojuca River, Pernambuco, Brazil. *Water Air Soil Pollution*, v. 180, p.261-269, 2007.  
 SUMNER, M.E. The electrical double layer and clay dispersion. In: SUMNER, M.E.; STEWART, B.A. (Eds.). *Soil crusting: chemical and physical processes*. Boca Raton: Lewis Publishers, 1992. p.1-31.  
 TAN, K.H. *Principles of soil chemistry*. 2ed. New York: Marcel Dekker, 1993. 362p.3