Langmuir Parameters and Soil Phosphorus Saturation Ratio

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Introduction

- Determination of Langmuir isotherms and subsequent calculation of the phosphorus (P) bonding strength (k), and the equilibrium P concentration (EPC_0) of a soil is a tedious and time-consuming process.
- Higher k-value indicates stronger bonding energy of P with Fe, AI oxides (i.e., low solution P).
- The P retention capacity of a non-calcareous soil depends on the Fe and Al content of the soil.
- The P saturation ratio (PSR) of a soil is the molar ratio of P to (Fe+AI).
- Based on a threshold PSR value (i.e., the PSR above which P release is high; Nair et al., 2004), the soil P storage capacity (SPSC) of a soil can be calculated (Nair and Harris, 2004).
- □ The SPSC, unlike the PSR, takes into account the Fe and AI responsible for the remaining soil P storage (see Eq. 1 under calculations).

SPSC_{Ox} = (0.05 – Soil PSR_{Ox})* [(Fe/56) + (Al/27)] * 31 (mg kg⁻¹)...Eq. 1 □ SPSC_{M1} = $(0.08 - \text{Soil PSR}_{M1})^*$ [(Fe/56) + (Al/27)] * 31 * 1.3 (mg kg⁻¹)Eq. 2

Calculations

 $[0.05 = \text{Threshold PSR}_{Ox}$ for Bh horizon and $0.08 = \text{Threshold PSR}_{M1}$ for Bh horizon of Spodosols; 95% confidence interval from 0 to 0.1 (Chakraborty et al., 2011) using SAS 9.2 software].

 \Box Adsorption parameters (k and EPC₀) were calculated using Langmuir equation:

$C/S = 1/k S_{max} + C/S_{max}$

Where, $S = S' + S_0$, the total amount of P sorbed, mg kg⁻¹

- S' = P sorbed by the solid phase, mg kg⁻¹
- $S_0 = Originally$ sorbed P on the solid phase, mg kg⁻¹
- C = Concentration of P after 24 h equilibration, mg L⁻¹

- □ Not able to obtain any specific value of k for PSR below the threshold (Fig. 3)
- □ Note: The threshold PSR for A horizons of Ultisols, Spodosols, Entisols and Alfisols is 0.10 (CI = 0.05 - 0.15; Nair et al., 2010) and the relationships noted here is expected to be applicable to all Florida A horizon soils.

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□ SPSC can be calculated from P, Fe and AI in an oxalate solution (Ox) or a soil test solution such as Mehlich 1 (M1) (Eq. 1 and Eq. 2) http://edis.ifas.ufl.edu/pdffiles/SS/SS54100.pdf

Hypothesis

- □ The Langmuir k will be high before the threshold PSR and will tend to zero once the threshold is reached.
- The k value will increase with increase in positive SPSC.
- \Box EPC₀ will be related to the PSR (and SPSC) as determined in an oxalate (Ox) solution or a soil test solution like Mehlich 1 (M1).

Objectives

- \Box To obtain a relationship between: i) EPC₀ and soil PSR, and ii) Langmuir k and soil PSR for Bh horizon of Spodosols dominated by Fe and Al.
 - Note: A and E horizons of Spodosols have no P sorbing capacity and generation of isotherms are not possible (Nair et al., 1998).
- To verify that the above k/PSR relationships are applicable to surface horizons of soils (A horizons of Ultisols taken as an example) where the Fe and AI content are lower.
- \Box Relate EPC₀ and Langmuir k to SPSC.

 $S_{max} = P$ sorption maximum, mg kg⁻¹ k = A constant related to bonding energy, L mg⁻¹ P Linearized form of equation: $S' = k'C - S_0$

 EPC_0 is the value of C when S' = 0



-600 -	•	$R^2 = 0.31$	
-800	EPC ₀ (μg mL- ¹ P)		

Fig. 4: Relationship between Soil P Storage Capacity (SPSC_{Ox}) calculated for the spodic horizon using P, Fe and AI in oxalate extract, and the equilibrium P concentration, EPC_0



Fig. 5: Relationship between Soil P Storage Capacity (SPSC_{Ox}) calculated for the spodic horizon using P, Fe and AI in oxalate extract, and the Langmuir k

 \Box EPC₀ is minimal below the threshold PSR; increases after the

Materials and Methods

Sampling locations: i) Lake Okeechobee Basin, Florida (Spodosols – sandy A and E horizons with no P retention capacity with an underlying spodic (Bh) horizon dominated by Fe and Al

ii) Suwannee River Basin, Florida (Ultisols -sandy marine sediment, A horizon)

Numbers of soil samples collected : 54 (Spodosols) and 10(Ultisols). **Soil Extractions & Chemical Analyses:**

✓ Soil pH

✓ Total elemental analysis using ICP

✓ Mehlich-1 extractable P, Fe, AI, Ca and Mg

- ✓ Oxalate extractable P, Fe and AI
- ✓ Water Soluble Phosphorus (WSP)
- ✓ Total Phosphorus (TP)
- Phosphorus Isotherm determination

References

□ Chakraborty D, V D Nair, M Chrysostome, and W G Harris, 2011. Soil phophorus storage capacity in manure-impacted Alaquods: Implications for water table management. Agric. Ecosyst. Environ.142: 167-175.

Fig. 1: Relationship between Water Soluble P (WSP) and P Saturation Ratio (PSR) for the spodic horizon calculated using P, Fe and AI in an oxalate extract (PSR_{ox}). Note: WSP increases rapidly after the threshold PSR of 0.05 (Charaborty et al, 2011).

Fig. 2: Relationship between Equilibrium P Concentration (EPC₀) and P Saturation Ratio (PSR) calculated for the spodic horizon using P, Fe and AI in oxalate extract (PSR_{ox}). Note that EPC_o shows an increase after the change point of 0.05.

 \Box WSP and EPC₀ minimal below the threshold PSR_{0x} (Figs. 1 and 2) \Box WSP increases with PSR_{Ox} above the threshold PSR_{Ox} (Fig. 1) \Box EPC₀ increases linearly (R² =0.72) with PSR (calculated in either oxalate or a soil test solution like Mehlich 1; only PSR_{Ox} shown here) (Fig. 2)



threshold value (Fig. 4)

- □ A SPSC/k relationship allows the strength of bonding (k) for soils below the threshold PSR (positive SPSC) to be estimated (Fig. 5).
- \Box Relationship between EPC₀/PSR and EPC₀/SPSC above threshold value might be better explained by other soil factors like pH, AI, Fe and Ca content of soils.

Multiple Regression Analysis for EPC_0 (above the threshold PSR): $EPC_0 = 13.37 - 4.40 \text{ pH}^* + 0.52 \text{ WSP}^* - 0.0009 \text{ TP} + 0.002 \text{ Ox-AI} +$ $0.001 \text{ Ox-Fe} - 0.001 \text{ Mg} + 0.012 \text{ Ca}^* (\text{R}^2 = 0.98, \text{P}=0.0025^*)$ \Box Empirical equation to predict EPC₀ from available soil characteristics using JMP Pro 9. \Box EPC₀ is positively related to WSP and M1-Ca but negatively related to pH.

Conclusions

- \Box EPC₀ and P bonding strength (k) can be estimated from P, Fe and AI in an oxalate or soil test solution like Mehlich 1 easily obtained from a soil testing lab without constructing Langmuir equations.
- □ P bonding strength increases with increase in positive SPSC (i.e., below the threshold soil PSR); such a relationship cannot be obtained from a k/PSR relationship which just indicates that k is high.
- □ The k/PSR relationship is applicable to surface horizons of all Florida

□ Nair V D, K M Portier, D A Graetz, and M L Walker, 2004. An environmental threshold for degree of phosphorus saturation in sandy soils. J. Environ. Qual. 33:107-113.

□ Nair V D and W G Harris, 2004. A capacity factor as an alternative to soil test phosphorus in phosphorus risk assessment. New Zealand J. Agric. Res. 4:491-497.

□ Nair V D, D A Graetz, and K R Reddy. 1998. Dairy manure influences on phosphorus retention capacity of Spodosols. J Environ Qual 27:522-527.

□ Nair VD, WG Harris and D Chakraborty . 2010. An indicator for risk of phosphorus loss from sandy soils . SL 333. http://edis.ifas.ufl.edu/pdffiles/SS/SS53900.pdf

Fig. 3: Relationship between P Saturation Ratio (PSR) calculated from the Bh horizon of Spodosols and A horizon of Ultisols using P, Fe and AI in an oxalate extract (PSR_{Ox}) and the P bonding strength (k) obtained from Langmuir isotherms. Threshold PSROx = 0.05 for Bh horizons in the graph and indicated with a read line Threshold PSROx = 0.10 for A horizons and indicated with a blue line

The PSR/k relationship suggests that the strength of P bonding is high below the threshold PSR (Fig. 3) for both surface horizon of Ultisols and Bh horizon of Spodosols.

soils where the Fe and AI content are lower.

□ Soils with different bonding energies (k) could potentially be used to rank the risk of P loss from land-use systems.

□ A simple procedure for determining k would be valuable when such values are needed as input in models for predicting P release from soils on a site-specific basis.

Acknowledgements

The authors would like to thank Debolina Chakraborty, Willie Harris and Debjani Sihi for their help and support during various stages of the work. We would also like to thank Dawn Lucas for her help with the analytical procedures.